



## Comprehensive metabolite and biological profile of “Sulmona Red Garlic” ecotype’s aerial bulbils

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### ABSTRACT

“Sulmona Red Garlic” is a well-known Italian traditional product. Bulbs, used for culinary purposes, have been largely investigated for their medicinal properties whereas aerial bulbils are usually removed as waste material. Here, for the first time, chemical composition and biological properties of the hydroalcoholic extract from aerial bulbils were investigated. Complementary information on metabolite composition were obtained using both NMR based untargeted and HPLC-DAD targeted methodologies. The NMR analysis revealed the presence of sugars, organic acids, amino acids, organosulphur compounds (methiin, alliin, allacin and cycloalliin), and other secondary metabolites. In particular, methiin and alliin were identified for the first time in the NMR spectra of aerial bulbil garlic extracts. Polyphenol content was determined by HPLC-DAD analysis: catechin, chlorogenic acid, and gallic acid turned out to be the most abundant phenolics.

Hydroalcoholic extract blocked cell proliferation of colon cancer cell line HCT116 with an IC<sub>50</sub> of 352.07 µg/mL, while it was non-toxic to myoblast cell line C2C12. In addition, it caused seedling germination reduction of two edible and herbaceous dicotyledon species, namely *Cichorium intybus* and *C. endivia*. Moreover, the same extract reduced the gene expression of TNF-α (tumor necrosis factor), HIF1-α (hypoxia-inducible factor), VEGFA (vascular endothelial growth factor), and transient receptor potential (TRP) M8 (TRPM8) indicating the ability to contrast cancer development through the angiogenic pathway. Final, *in silico* experiments were also carried out supporting the biological effects of organosulphur compounds, particularly alliin, which may directly interact with TRPM8.

The results here reported suggest the potential use of garlic aerial bulbils often considered a waste product as a source in phytotherapeutic remedies.

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## 1. Introduction

Garlic (*Allium sativum* L.) a member of the Amaryllidaceae family, has been used worldwide for culinary and medicinal purposes since ancient times. Bulbs are the most commonly used part of the plant; leaves and flowers are also edible but their consumption is limited as they are rather considered waste products.

Garlic aerial flowers, known as bulbils, are small bulb-like structures that form above the ground on the flower stalks of garlic plants. These bulbils are reproductive structures that contain embryonic garlic plants and can develop when garlic plants are allowed to flower, as they are a mean of propagation for the plant. However, cultivated garlic is usually grown from seed cloves, and aerial bulbils are typically removed early on to prevent flowering and promote bulb growth (Kamenetsky et al., 2004). These aerial bulbils can be used as a local food resource or for biotechnology purposes, making them potentially interesting by-products (Dong et al., 2022; Kajimura et al., 2000; Mathew et al., 2005). Literature studies report generally the phytochemical profile and the biological properties of garlic bulbs and leaves (Kasprzak-Drozd et al., 2022; Masashi et al., 2020; Snirc et al., 2023) whereas aerial bulbils have not yet been investigated.

Among the various garlic ecotypes, “Sulmona Red Garlic” is a well-known Italian traditional product listed in the “National Register of Horticultural Varieties” (DM 296/2009 delivered by the Italian Ministry of Agricultural, Food and Forestry Policy; “Sulmona Red Garlic”, Ark of taste, Slow Food Foundation for Biodiversity). It has been grown for a long time in the Abruzzo region, specifically in the surroundings of Sulmona. Bulbs are characterized by a mottled wine-red tunic, whereas, according to the production regulation, the aerial bulbils are removed, being a waste material of this food chain. However, the careful and manual work underlying the selection of the bulbils in the productive chain makes this by-product a high-quality plant material that is also considered a second line-line commercial product (La Salvia et al., 2022). Previous pharmacological studies conducted on the bulbs of “Sulmona red garlic” demonstrated antioxidant and anti-inflammatory effects in brain synaptosomes and monocytes, respectively (Brunetti et al., 2009; Lasalvia et al., 2022). The chemical profile of bulbs from Sulmona Red Garlic have been also investigated using different analytical techniques (Ritota et al., 2012; Ferioli et al., 2020; La Salvia et al., 2011; La Salvia et al., 2020; Biancolillo et al., 2020).

In these investigations, the organo-sulphur compounds and other metabolome components, such as amino acids, fatty acids, organic acids and sugars, were recognised as promising quality and traceability indicators to assess the garlic provenance or variety (Ritota et al., 2012; Biancolillo et al., 2020), to assess the efficiency of different extraction methods (Ferioli et al., 2020), and to correlate the chemical composition to the biological properties (La Salvia et al., 2022) including cardioprotective (Recinella et al., 2021) and cholesterol-lowering effects (Liu et al., 2000), inhibiting carcinogenesis and cancer growth (Shirin et al., 2001; Yan et al., 2013), and enhancing immunocompetence (Xu et al., 2018; Nantz et al., 2012).

Recently, the interest in garlic’s potential to protect against inflammatory bowel diseases (IBDs) has significantly grown. Polar extracts from Nubia ecotype bulbs have shown an effect in reducing the burden of inflammation and oxidative stress, in isolated mouse colon (Recinella et al., 2022). Studies on rats have shown that garlic oil, which contains beneficial organosulphur compounds like allicin, can help reduce tissue damage caused by inflammation in an experimental in vivo model of colitis (Tanrikulu, Şen Tanrikulu, Kılınç, Can, & Köktürk, 2020). Additionally, clinical studies have pointed to a reduction of colon adenomas through the modulation of natural killer cell activity (Ansary et al., 2020). These effects have been associated to the presence in bulbils of specialized metabolites, such as sulphur and phenolic compounds (Kopeć et al., 2020). Sulphur compounds have shown protective effects in the colon, in terms of both anti-inflammatory and chemopreventive effects (Saud et al., 2016; Vezza et al., 2019). Similarly, phenolic

compounds have long been involved in inhibiting the production of inflammatory mediators, in isolated colon exposed to inflammatory stimulus (Ferrante et al., 2019; Menghini et al., 2016; Menghini et al., 2018; Recinella et al., 2022).

In the present study, for the first time, the hydroalcoholic extracts of aerial bulbils of “Sulmona Red Garlic” have been investigated to obtain the chemical composition and biological properties. The phytochemical composition has been obtained by applying both untargeted and targeted analytical methodologies. In particular, untargeted Nuclear Magnetic Resonance (NMR) was used to obtain a comprehensive metabolite profile, allowing the identification and quantification of many compounds at once (Mannina, Sobolev, & Capitani, 2012; Mannina, Sobolev, & Viel, 2012). Targeted HPLC-DAD was used to obtain the phenolic profile. Cytotoxic effects of the bulbil extract were studied in the human colon cancer (HCT116) cell line, in which the gene expression of transient receptor potential (TRP) M8 (TRPM8), possibly involved in colon carcinogenesis (Borrelli et al., 2014), was measured. Additionally, the gene expression of inflammatory and angiogenic mediators such as tumor necrosis factor (TNF) $\alpha$ , hypoxia-inducible factor (HIF)1 $\alpha$ , and vascular endothelial growth factor (VEGF)A was evaluated. Finally, *in silico* experiments, including components-targets analysis and docking runs, were conducted for unravelling the putative mechanisms underlying the observed biological effects.

## 2. Materials and Methods

### 2.1. Chemicals

Double-distilled water was obtained using a Millipore Milli-Q Plus water treatment system (Millipore Bedford Corp., Bedford, MA, USA). EDTA deuterated was purchased from Cambridge Isotope laboratories, Inc. (Andover, USA). Monobasic potassium phosphate (KH<sub>2</sub>PO<sub>4</sub>) and dibasic potassium phosphate (K<sub>2</sub>HPO<sub>4</sub>) were purchased from Sigma-Aldrich S.r.l. (Milan, Italy). Deuterated water (D<sub>2</sub>O) 99.97 atom% of deuterium, methanol-*d*<sub>4</sub> 99.80 atom% of deuterium. 3-(Trimethylsilyl)propionic-*d*<sub>2,2,3,3</sub>-*d*<sub>4</sub> acid sodium salt (TSP) was purchased from Merck (Milan, Italy).

### 2.2. Sampling and sample preparation to perform HPLC analysis and biological assays

Plant material was sampled during the transformation process, from a commercial batch produced by “Aglia D’Alessandro” company (Pratola Peligna, L’Aquila, Italy). The plant material was a kind gift from “Aglia D’Alessandro” company. According to the “Sulmona Red Garlic” production regulation (“Sulmona Red Garlic”, Ark of taste, Slow Food Foundation for Biodiversity. <https://www.fondazione Slow Food.com/en/ark-of-taste-slow-food/sulmona-red-garlic/>), the floral buds were removed at early stage of the emission. In late June, the whole plants were collected from crops and left to gently dry in a ventilated area, without direct sun exposure. In the following weeks, plants were carefully checked to remove unhealthy bulbs and prepared for commercialization that is in the form of whole bulbs, after removal of roots, leaves, and the outer layer of tunica, or prepared in the form of the typical garlic braids. In July 2021, the aerial bulbils (Supplementary Material) were manually isolated from the commercial batch and collected for the experiments. The bulbils belong to a representative sample of 4 Kg small size bulbils.

### 2.3. Extraction procedure for NMR analysis

The plant material was freeze dried in bulk for 3 days by Buchi Lyovapor 1–200 at –55 °C and 0.200 mbar until complete loss of water. Each sample was homogenized using a knife mill and was immediately subjected to Bligh-Dyer extraction protocol (Bligh and Dyer, 1959). In particular, the lyophilized sample (0.1 g) was added with 3 mL of a

**Table 1**

Metabolites identified in the 600.13 MHz <sup>1</sup>H NMR spectrum of the hydroalcoholic Bligh-Dyer extract of “Sulmona Red Garlic” aerial bulbils. Asterisks indicate signals used for the integration and quantification of metabolites.

Compound	Assignment	<sup>1</sup> H (ppm)	Multiplicity (J (Hz))	<sup>13</sup> C (ppm)
<i>Sugars</i>				
α-D-Fructofuranose	CH-3	4.11		83.2
β-D-Fructofuranose	CH-4	4.12		75.6
	CH-5	3.86		81.9
α-Glucose (α-Glc)	CH-1	5.24*	d [4.0]	91.1
	CH-2	3.56		72.1
	CH-4	3.41		70.1
	CH-5	3.81		73.3
β-Glucose (β-Glc)	CH-1	4.65*	d [8.0]	96.8
	CH-2	3.28		74.9
	CH-5	3.48		76.7
	CH <sub>2</sub> -6	3.73; 3.88		61.5
Sucrose (Suc)	CH-1 (Glc)	5.42*	d [3.8]	93.2
	CH-2	3.57		72.0
	CH-3	3.72		73.6
	CH-4	3.48		70.0
	CH-5	3.82		73.4
	CH-3'(Fru)	4.23	d [8.6]	77.8
	CH-5'	3.88		82.3
	CH-4'	3.83		61.3
	CH <sub>2</sub> -1'	3.69		59.7
<i>Organic Acids</i>				
Acetic acid (AA)	α-CH <sub>3</sub>	1.92*	s	24.6
Citric acid (CA)	α,γ-CH	2.56	d [15.1]	45.9
	α',γ'-CH	2.68*	d [15.1]	45.9
Formic acid (FA)	HCOOH	8.45*	s	
Fumaric acid (FumA)	α, β-CH = CH	6.52*	s	
Malic acid (MA)	α-CH	4.31*	dd [10.1; 3.1]	71.3
	β-CH	2.68	dd [15.5; 3.1]	43.6
	β'-CH	2.38		43.6
Pyruvic acid (PA)	CH <sub>3</sub>	2.40*	s	30.6
<i>Amino acids</i>				
Alanine (Ala)	β-CH <sub>3</sub>	1.49*	d [7.2]	17.1
	α-CH	3.80		51.3
Arginine (Arg)	α-CH	3.80		55.1
	β-CH <sub>2</sub>	1.93	m	28.5
	γ, γ'-CH	1.73; 1.69*	m	24.9
Asparagine (Asn)	δ-CH <sub>2</sub>	3.26		41.4
	α-CH	4.04		52.1
	β-CH	2.87*	dd [7.8; 16.9]	35.5
	β'-CH	2.98		35.5
γ-amino butyric acid (GABA)	α-CH <sub>2</sub>	2.30*	t [7.4]	35.4
Glutamate (Glu)	γ-CH <sub>2</sub>	3.03	t	40.0
	γ-CH <sub>2</sub>	2.35	m	34.4
	β, β'-CH <sub>2</sub>	2.07*, 2.13		29.9
Glutamine (Gln)	α-CH	3.79		55.1
	α-CH	3.80		55.3
	β-CH <sub>2</sub>	2.15	m	27.0
	γ-CH <sub>2</sub>	2.46*	m	32.6
Glycine	α-CH <sub>2</sub>	3.57*	s	42.2
Histidine (His)	CH-5	7.17	s	117.8
	CH-2	8.07*	s	136.5
Isoleucine (Ile)	γ-CH <sub>3</sub>	1.01*	d [7.1]	17.4
	δ-CH <sub>3</sub>	0.89		
Leucine (Leu)	β-CH <sub>2</sub>	1.75	m	
	δ-CH <sub>3</sub>	0.96*	d [6.2]	21.7
	δ'-CH <sub>3</sub>	0.97	d [6.2]	
Lysine (Lys)	α-CH	3.77		55.2
	β-CH <sub>2</sub>	1.91	m	28.5
	ε-CH <sub>2</sub>	3.04*	t [7.4]	40.0
Phenylalanine (Phe)	CH-2,6	7.30		130.1
	CH-4	7.37*		129.6
	CH-3,5	7.43		
Proline (Pro)	α-CH	4.15		61.7
	β, β'-CH <sub>2</sub>	2.07, 2.37		30.0
	γ-CH <sub>2</sub>	2.01*	m	24.7

**Table 1 (continued)**

Compound	Assignment	<sup>1</sup> H (ppm)	Multiplicity (J (Hz))	<sup>13</sup> C (ppm)
	δ, δ'-CH <sub>2</sub>	3.35, 3.40		47.1
Threonine (Thr)	γ-CH <sub>3</sub>	1.33*	d [6.7]	20.2
Tryptophan (Trp)	CH-7	7.53	d [8.1]	113.2
	CH-4	7.73*	d [8.1]	119.4
Tyrosine (Tyr)	CH-2,6	7.17	d [8.6]	131.3
	CH-3,5	6.85*	d [8.6]	116.6
Valine (Val)	γ-CH <sub>3</sub>	0.90	d [7.1]	17.6
	γ'-CH <sub>3</sub>	1.05*	d [7.1]	18.9
<i>Organosulphur compounds</i>				
Alliin	CH <sub>2</sub> -6	5.51	dq [1.2;17.0]	126.4
		5.57	<sup>trans</sup> dq [1.2; 10.2] <sup>cis</sup>	126.4
	O = S-CH <sub>2</sub> -4	3.64; 3.87	dd	55.9
	CH <sub>2</sub> -3	3.22; 3.44		50.9
	CH-5	5.94	m	125.5
Alliin	CH <sub>2</sub> -6	5.51	dq [1.2;17.0]	126.4
		5.57	<sup>trans</sup> dq [1.2; 10.2] <sup>cis</sup>	126.4
	CH <sub>2</sub> -1	5.17	dq [1.6; 10.0] <sup>cis</sup>	118.9
		5.21	dq [1.6; 18.6]	118.9
	O = S-CH <sub>2</sub> -4	3.64; 3.87	<sup>trans</sup> dd	55.9
	CH-2	5.84*	m	134.7
	CH-5	5.94	m	125.5
	S-CH <sub>2</sub> -3	3.24	dd	41.3
Cycloalliin	CH <sub>3</sub>	1.44*	d [6.5]	19.5
	O = S-CH <sub>2</sub> -6	2.81; 3.23		47.7
	N-CH-5	3.93		43.7
S-Methyl-L-cysteine-S-oxide (Methiin, MCSO)	γ-CH <sub>3</sub>	2.83*	s	39.1
	β-CH <sub>2</sub>	3.27, 3.47		54.0
	α-CH	4.18		59.3
Sulphur compounds (OSCs)	CH <sub>2</sub> = CH-	5.99	dq	122.3
<i>Miscellaneous</i>				
Choline	N(CH <sub>3</sub> ) <sub>3</sub>	3.20*	s	54.9
Glycine betaine	N(CH <sub>3</sub> ) <sub>3</sub>	3.27*	s	54.4
	α-CH <sub>2</sub>	3.81		67.3
Pyroglutamic acid	β, β'-CH <sub>2</sub>	2.04, 2.51*	m	26.5
		2.40		
	γ-CH <sub>2</sub>	2.40		32.5
Trigonelline	CH-6	9.13*	s	
	CH-4,2	8.84		
Tyramine	CH-3,5	7.21		131.6
	CH-2,6	6.90*	d	116.6
Uridine	CH-1' (rib)	5.90		
	CH-6	7.88*	d [7.9]	

d = doublet; dd = doublet of doublets; dq = doublet of quartet; m = multiplet; s = singlet; t = triplet.

CH<sub>3</sub>OH/CHCl<sub>3</sub> mixture (2:1 v/v) and 0.8 mL of double-distilled water. The resulting system was sonicated at room temperature for 10 min, followed by the addition of 1 mL of chloroform and 1 mL of double-distilled water. The hydroalcoholic and organic phases were finally separated after centrifugation (Eppendorf Centrifuge 5430 R (Milan, Italy)) for 15 min (25 °C, 7540 g). The residual pellets were extracted two more times under the same conditions as described. The three extractions obtained from each step were combined. The hydroalcoholic fraction were dried using nitrogen flow. The dried extracts were stored at - 20 °C until further analysis.

#### 2.4. NMR analysis

NMR analyses were carried out on a Jeol JNM-ECZ 600R operating at the proton frequency of 600.17 MHz and equipped with a Jeol 5 mm FG/RO DIGITAL AUTOTUNE probe. <sup>1</sup>H NMR experiments were carried out

**Table 2**

Gradient elution condition of HPLC-DAD analyses of the hydroalcoholic extract of "Sulmona Red Garlic" bulbs.

TIME (min)	COMPOSITION A% (Water + Formic acid 0.1 %)	COMPOSITION B% (Methanol + Formic acid 0.1 %)	FLOW (mL/min)
1	97	3	0.6
5	77	23	0.6
12	73	27	0.6
18	57	43	0.6
25	52	48	0.6
32	50	50	0.6
34	50	50	0.6
37	35	65	0.6
40	5	95	0.6
47	10	90	0.6
48	10	90	0.6

by using the following parameters: 298 K, 128 scans, residual water signal suppression with a presaturation pulse, 7.73 s relaxation delay, 90° pulse of 8.3 μs, 64 k data points, and 9000 Hz spectral width. Homonuclear <sup>1</sup>H-<sup>1</sup>H TOCSY experiment was acquired with 64 scans, 8 k data points in *f*<sub>2</sub> and 128 in *f*<sub>1</sub>, 50 ms mixing time, 2 s relaxation delay, and 9000 Hz spectral width in both dimensions. Heteronuclear <sup>1</sup>H-<sup>13</sup>C HSQC experiment was acquired with 64 scans, 8 k data points in *f*<sub>2</sub> and 256 in *f*<sub>1</sub>, 2 s relaxation delay, and a spectral width of 9000 Hz and 33000 Hz for *f*<sub>2</sub> and *f*<sub>1</sub>, respectively (Spano et al., 2023) (Spano, M.; Di Matteo, G.; Fernandez Retamozo, C.A.; Lasalvia, A.; Ruggeri, M.; Sandri, G.; Cordeiro, C.; Sousa Silva, M.; Totaro Fila, C.; Garzoli, S.; et al. A Multimethodological Approach for the Chemical Characterization of Edible Insects: The Case Study of *Acheta domesticus*. *Foods* 2023, 12, 2331. <https://doi.org/10.3390/foods12122331>).

The dried hydroalcoholic phase was dissolved in 700 μL of 100 mM phosphate buffer/D<sub>2</sub>O, containing 0.4 mM TSP as internal standard. <sup>1</sup>H spectra were referenced to methyl group signals of TSP (δ<sub>H</sub> = 0.00 ppm) in D<sub>2</sub>O. Spectra processing, and signals integration were carried out with JEOL Delta software (v5.3.1).

Metabolite quantification was carried out by integration of selected signals (marked with asterisk) listed in Table 1 using JEOL Delta software (v5.3.1) and normalized with respect to the methyl group signal of TSP (0.00 ppm), set to 100. The quantified metabolites were expressed as mg/100 g of dried sample. NMR chemical shift used for signal integration are reported with an asterisk in Table 1.

## 2.5. Extraction procedure for phenolic HPLC analysis and biological studies

Extracts for phenolic analysis and biological studies were prepared according to a protocol already described (Lasalvia et al., 2022; Recinella et al., 2022). In particular, the extraction was performed by maceration in a sonicator bath at a frequency of 30 kHz [2 consecutive extractions of 30 min each at room temperature (20 °C)] with water-ethanol (50:50, v/v). The plant material/solvent ratio was 1/20 (w/w). The extract was freshly prepared and centrifuged at 5000g (Menghini et al., 2018).

## 2.6. HPLC determination of phenolic compounds

The quantitative determination of phenols present in the water-ethanol (50:50, v/v) extract was carried out using a reversed-phase HPLC-DAD in gradient elution mode as previously described (Acquaviva et al., 2022). Phenols separation was conducted within 60 min of the chromatographic run, starting from the following separation conditions: 97 % water with 0.1 % formic acid, 3 % methanol with 0.1 % formic acid (Table 2). The separation was performed on an Infinity lab Poroshell 120-SB reverse phase column (C18, 150 × 4.6 mm i.d., 2.7 μm; Agilent, Santa Clara, CA, USA). Column temperature was set at 30 °C.

Quantitative determination of phenolic compounds was performed via a DAD detector. Quantification was done through 7-point calibration curves, with linearity coefficients (R<sup>2</sup>) > 0.999, in the concentration range 2–140 μg/mL. All standards were purchased from Sigma Aldrich (Milan, Italy), and have a purity ≥ 95 %. The limits of detection were lower than 1 μg/mL for all assayed analytes. The area under the curve from HPLC chromatograms was used to quantify the analyte concentrations in the extract.

## 2.7. Eco-toxicological studies

Allelopathy bioassay was carried on the seeds of the herbaceous dicotyledon species *Cichorium intybus* and *C. endivia* because of their fast germination rate and high sensitivity. The detailed procedure was conducted as previously reported (Ferrante et al., 2019). Seeds were treated with scalar *A. sativum* extract concentrations (0.625–40 mg/mL) and considered germinated for observed root length ≥ 1 mm, after the third day of treatment. *Artemia salina* (brine shrimp) was cultured as previously described (Orlando et al., 2021), and toxicity induced by the extract (0.1–20 mg/mL) was expressed in terms of LC<sub>50</sub> values. *Daphnia magna* were cultured as previously described (Villegas-Navarro, Rosas-L., & Reyes, 2003). Briefly, non-pregnant *Daphnia magna* were maintained separately in 50 mL of extract solution (10 mg/mL) at room temperature throughout the experiment. The hearth rate was recorded through microscopy. *Daphnia magna* were placed individually onto a single cavity microscope slide in a 50 μL droplet of the tested extract, where the heartbeat rate was measured three times for 10 s. Each measurement was conducted in triplicate. The decrease in hearth rate was compared to both untreated and 30 % ethanol-treated groups, working as negative and positive controls, respectively.

## 2.8. In vitro study

### 2.8.1. Human cardiomyocyte C2C12 cell line

Human cardiomyocyte C2C12 cell line was cultured in DMEM (Euroclone) supplemented with 10 % (v/v) heat-inactivated fetal bovine serum and 1.2 % (v/v) penicillin G/streptomycin in 75 cm<sup>2</sup> tissue culture flask (n = 5 individual culture flasks for each condition). The cultured cells were maintained in humidified incubator with 5 % CO<sub>2</sub> at 37 °C. For cell differentiation, C2C12 cell suspension at a density of 1 × 10<sup>6</sup> cells/mL was treated with various concentrations (10, 50, and 100 ng/mL) of phorbol myristate acetate (PMA, Fluka) for 24 h or 48 h (induction phase). Thereafter, the PMA-treated cells were washed twice with ice-cold pH 7.4 phosphate buffer solution (PBS) to remove PMA and non-adherent cells, whereas the adherent cells were further maintained for 48 h (recovery phase). Morphology of cells was examined under an inverted phase-contrast microscope. To assess the basal cytotoxicity of the extract, a viability test was performed on 96 microwell plates, using 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) test. Cells were incubated with the extract (ranging in the concentration 10–1000 μg/mL) for 24 h. 10 μL of MTT (5 mg/mL) was added to each well and incubated for 3 h. The formazan dye formed was extracted with dimethyl sulfoxide and absorbance was recorded as previously described (Menghini et al., 2018). Effects on cell viability was evaluated in comparison to untreated control group (Ctrl).

### 2.8.2. Human colon cancer-derived HCT116 cell

Human colon cancer-derived HCT116 cell were cultured in DMEM (Euroclone) supplemented with 10 % (v/v) heat-inactivated fetal bovine serum and 1 % (v/v) penicillin G/streptomycin in a 75 cm<sup>2</sup> cell culture flasks. The cultured cells were maintained in a humidified incubator with 5 % CO<sub>2</sub> at 37 °C. When the confluency reached 80 %, a viability test was performed using 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) test to assess the basal cytotoxicity of the extract under investigation. For this assay cells were seeded (5 × 10<sup>3</sup> cells/well) onto flat-bottomed 96-well culture plates and incubated

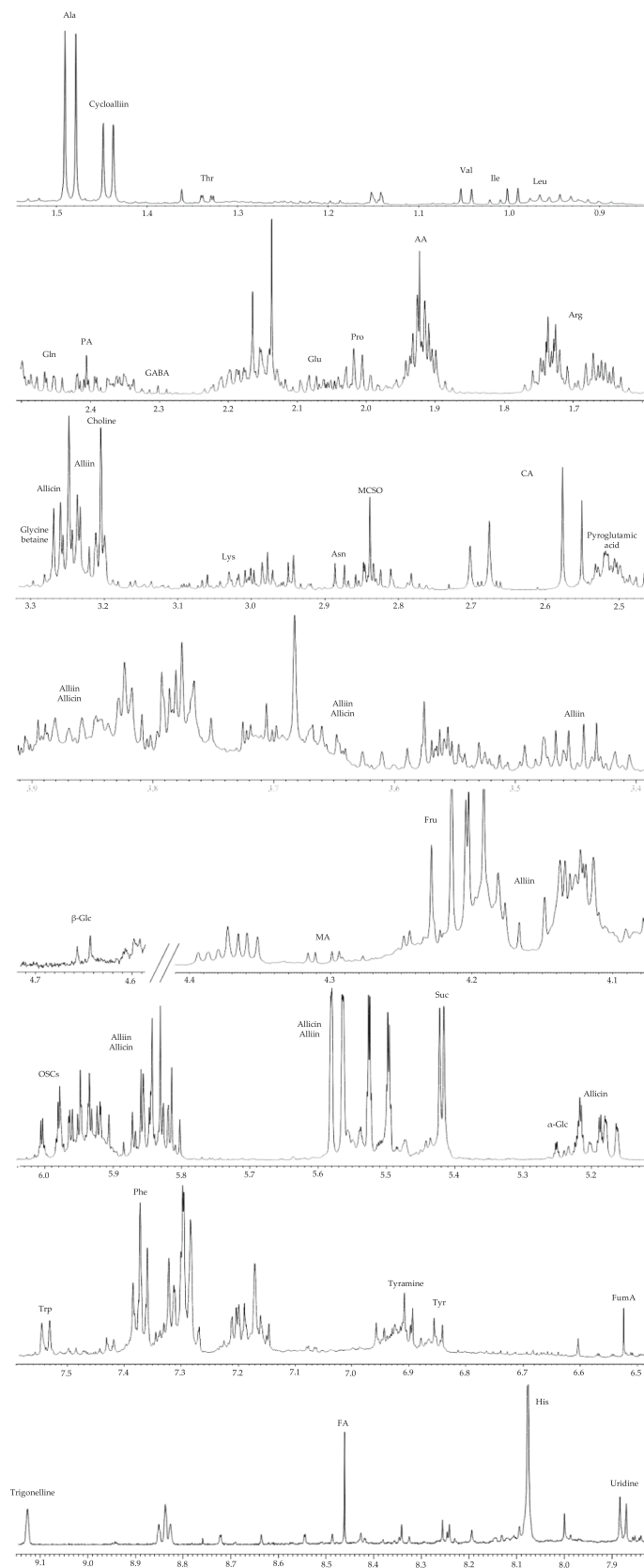
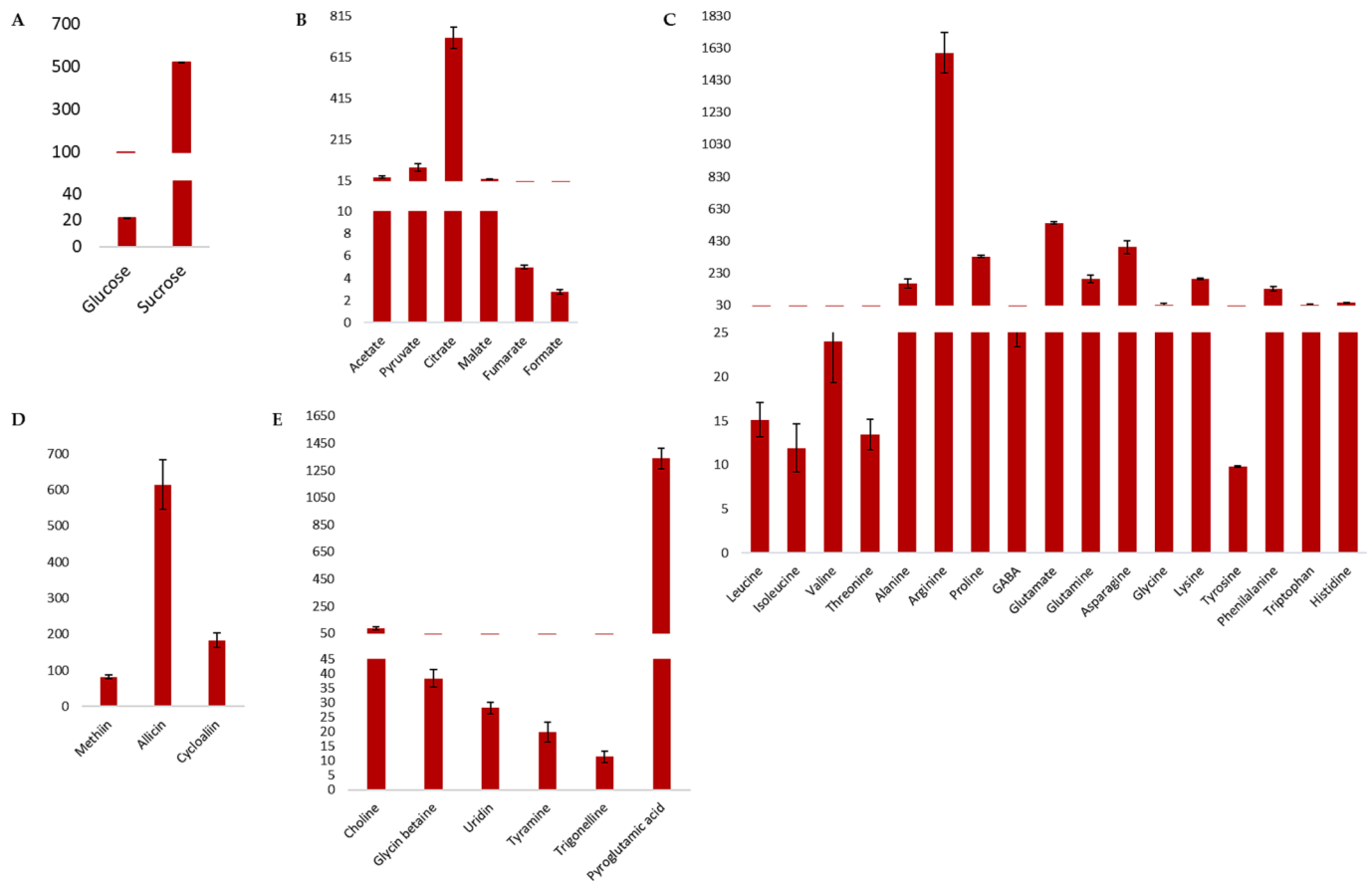


Fig. 1. <sup>1</sup>H NMR spectrum of the hydroalcoholic Bligh-Dyer extract, in 100 mM PBS/D<sub>2</sub>O, 0.4 mM TSP.



**Fig. 2.** Histograms resulting from the quantitative NMR analysis of quantified compounds (mg/100 mg dry weight) present in the Blich-Dyer hydroalcoholic extracts of red garlic aerial bulbils. (A) Sugars, (B) Organic acids, (C) Amino acids, (D) Organosulphur compounds, and (E) Miscellaneous. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

overnight. After 24 h, the extract at different concentrations (200–1000  $\mu\text{g}/\text{mL}$ ) was added and the plates were incubated for 24 h. After this time, a total of 10  $\mu\text{L}$  of MTT (5 mg/mL in PBS) was added to each well and incubated for 3 h. The formazan dye formed was solubilized with dimethyl sulfoxide and absorbance was recorded at 540 nm. Effects on cell viability were evaluated in comparison to the control (Ctrl: untreated cells) and expressed as a percentage of the control culture value. In a second set of experiments cells were treated with either vehicle or extract for 24 h and subsequently post-treated with hydrogen peroxide (H.P.) 300  $\mu\text{M}$  for 3 h. After this time, cell survival was determined by MTT assay as described above. Each condition was run in triplicate, including untreated control and blank cell-free control.

Total RNA was extracted from HCT116 cells using TRI reagent (Sigma-Aldrich, St. Louis, MO, USA), according to the manufacturer's protocol, and reverse transcribed using High-Capacity cDNA Reverse Transcription Kit (Thermo Fischer Scientific, Waltman, Massachusetts, USA). Gene expression of TNF $\alpha$ , TRPM8, HIF1 $\alpha$ , and VEGFA were determined by quantitative real-time PCR using TaqMan probe-based chemistry, as previously described (Menghini et al., 2018). PCR primers and TaqMan probes were purchased from Thermo Fisher Scientific Inc.  $\beta$ -actin was used as the house-keeping gene. The elaboration of data was conducted with the Sequence Detection System (SDS) software version 2.3 (Thermo Fischer Scientific). Relative quantification of gene expression was performed by the comparative  $2^{-\Delta\Delta\text{Ct}}$  method (Livak & Schmittgen, 2001).

## 2.9. In silico predictions and molecular modelling

The putative targets and pathways underlying the mechanism of

action of the extract were predicted through the bioinformatics platform STITCH (accessed on 11th April 2023), as previously described (Ferrante et al., 2020; Gu et al., 2020).

The cryo-EM structure of the TRPM8 ion channel complexed with icilin was retrieved from the protein data bank (PDB ID: 6NR3) (<https://www.rcsb.org/>) (accessed on 18th February 2023). The protein was prepared using Playmolecule web server (<https://playmolecule.com/proteinPrepare/>) (accessed on 18th February 2023) (Martínez-Rosell, Giorgino, & De Fabritiis, 2017). The 3D structure of alliin was retrieved from the PubChem database (<https://pubchem.ncbi.nlm.nih.gov/>) (accessed on 18th February 2022) and its geometry was optimized using Biovia Discovery Studio Visualizer v4.5 (Dassault Systèmes Biovia Software Inc, San Diego, CA, USA, 2012). The binding coordinates of the cocrystal ligand icilin were used to define the docking grid box x, y, z dimension (50, 50, 50 Å) and x, y, z center coordinates (171.7, 137.5, 174.9 Å) with 0.375 Å using AutoDockTools 1.5.6. Docking was carried out using AutoDock 4.2.6 (<https://autodock.scripps.edu>) (Morris et al., 2009). The docking protocol, including the algorithm and the number of energy evaluations, have been previously described (Albayati, Uba, & Yelekcı, 2022; Llorent-Martínez et al., 2022). The docking score of each ligand pose was calculated, and protein–ligand interaction was analyzed using Biovia Discovery Studio Visualizer v4.5 (Dassault Systèmes Biovia Software Inc, San Diego, CA, USA, 2012).

## 2.10. Statistical analysis

The software GraphPad Prism version 5.01 (Graphpad Software Inc., San Diego, CA, USA) was used to perform data analysis. Means  $\pm$  SD

were determined for each experimental group and analyzed by ANOVA, followed by Newman-Keuls multiple comparison post hoc test. The limit of statistically significant differences between mean values was set at  $p$ -value  $< 0.05$ .

### 3. Results and discussion

#### 3.1. 3.1 Metabolomic profile of bulbils aqueous extract by NMR analysis

The  $^1\text{H}$  spectrum of Bligh-Dyer hydroalcoholic extracts from “Sulmona Red Garlic” bulbils, see Fig. 1, was assigned using 2D NMR experiments ( $^1\text{H}$ - $^1\text{H}$  TOCSY,  $^1\text{H}$ - $^{13}\text{C}$  HSQC) and literature data on other vegetable food matrices (Ingallina et al., 2023; Liang et al., 2015; Ritota et al., 2012; Saviano et al., 2019; Spano et al., 2021). In some specific cases, the comparison with the NMR spectra of pure reference standard were also carried out to confirm the compound assignment.

In the high-field 0.8–3.6 ppm spectral region, signals of methyl and methylene groups belonging to aliphatic amino acids and organic acids are observed. The spectral region between 3.0 ppm 5.5 ppm is dominated by very intense signals of principal monosaccharides and disaccharides, and organosulphur compounds.

In the low frequency spectral region (6.0–9.0 ppm),  $^1\text{H}$  signals of amino acid aromatic groups (phenylalanine, tryptophan, and tyrosine), formic and fumaric acids and trigonelline, uridine and tyramine were detected.

Hereunder, each class of compounds is discussed separately.

##### 3.1.1. Sugars

The  $^1\text{H}$  NMR spectrum revealed the presence of mono- and disaccharides characterized by distinct  $^1\text{H}$  resonances due to anomeric CH-1 group:  $\beta$ -D-glucose and  $\alpha$ -D-glucose signals at 5.24 ppm and 4.65 ppm, sucrose and fructose signal at 5.42 ppm and 4.11 ppm, respectively. Sucrose and glucose were quantified showing a content of  $518.98 \pm 0.89$  and  $21.50 \pm 0.29$  mg, respectively (Fig. 2A).

This class of metabolites play a key role in the physiological activities of bulbils over to ensure the basal metabolism of autotrophic cells. In garlic, and generally in plants, sugars act as a carbohydrate reserve and soluble molecules able to modulate osmoregulation and prevent cell damage. They are usually mobilized and used from plant during growth phases and sprout development.

##### 3.1.2. Organic acids

$^1\text{H}$  NMR spectrum showed the presence of acetic, pyruvic, citric, malic, fumaric and formic acids. The presence of acetic acid was confirmed by the methyl group singlet at 1.92 ppm and the relative correlation with carbon atom at 24.6 ppm in the  $^1\text{H}$ - $^{13}\text{C}$  HSQC spectrum.

The singlet at 2.40 ppm which correlate with carbon at 30.6 ppm referred to methyl group of pyruvic acid. Citric acid was identified by the typical doublets ( $J_{\text{H-H}} = 15.1$  Hz) at 2.56 and 2.68 ppm of the symmetric diastereotopic  $\text{CH}_2$  groups that correlated with carbon at 45.9 ppm in  $^1\text{H}$ - $^{13}\text{C}$  HSQC spectrum. Citrate is a clavation products of amino acids and, with sulfur compounds, allows to define the sensory profile of bulbs and bulbils (Jones et al., 2004).

According with literature (Liang et al., 2015; Ritota et al., 2012) also malic acid was found due to the presence of double doublets ( $J_{\text{H-H}} = 10.1$  and 3.1 Hz) at 4.31 ppm corresponding to the  $\alpha$ -CH proton. The cross-peaks in  $^1\text{H}$ - $^1\text{H}$  TOCSY experiment showed the mutual correlation with  $\beta$ -CH (2.68 ppm) and  $\beta'$ -CH (2.38 ppm) protons. Moreover, the  $^1\text{H}$ - $^{13}\text{C}$  HSQC spectrum confirmed the short-range correlation between each proton with corresponding carbons at 71.3 and 43.6 ppm, respectively. Malate could not be quantified by the integration of  $\beta$ - $\text{CH}_2$  doublets due to their poor resolution being overlapped with one doublet (2.68 ppm) of citrate, thus the signal at 4.31 ppm had been chosen for integration. Both fumaric and formic acids were identified by the presence of the typical singlets at 6.52 ppm and 8.45 ppm, respectively. The most abundant organic acid turned out to be citrate (over 700 mg/100 g

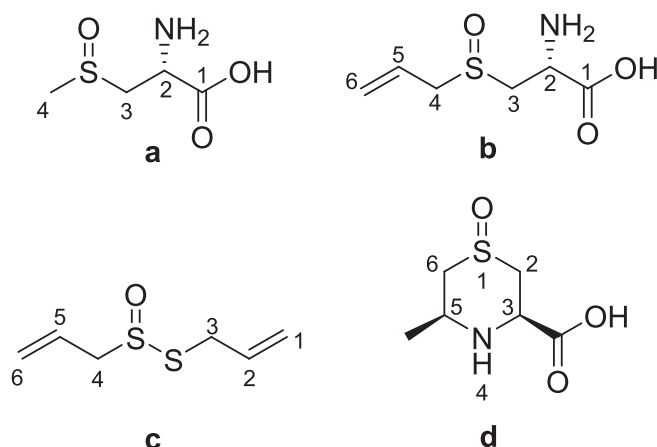


Fig. 3. Structures of organosulphur compounds: (a) methiin; (b) alliin; (c) allicin; (d) cycloalliin.

DW), followed by pyruvate, acetate, and malate, above 100 mg/100 g DW (Fig. 2B). Fumaric and formic acids were found in very low concentrations, below 5 mg/100 g dry weight matrix.

**3.1.2.1. Amino acids.** Seventeen amino acids, namely alanine, arginine, asparagine, GABA, glutamic acid, glutamine, glycine, histidine, isoleucine, leucine, lysine, phenylalanine, proline, threonine, tyrosine, tryptophan, and valine were identified in the  $^1\text{H}$  spectrum. In addition to GABA, two other non-protein amino acids, namely methiin and alliin, typically found in vegetables of Brassicaceae and Liliaceae (Amaryllidaceae) family, were also identified. Their assignment will be discussed in (3.1.4) section together with the other organosulfur compounds.

Signals in the range from 0.90 to 1.10 ppm arise from methyl or methylene groups of valine, leucine, and isoleucine,  $^1\text{H}$ - $^1\text{H}$  and TOCSY correlations allowed the correct assignment. The multiplet at 1.69 ppm and its correlation with the carbon at 24.9 ppm in the  $^1\text{H}$ - $^{13}\text{C}$  HSQC represented the typical spin pattern of  $\gamma$ -CH of arginine.  $\alpha$ -CH ( $^1\text{H}$  3.80 and  $^{13}\text{C}$  55.1 ppm),  $\beta$ - $\text{CH}_2$  ( $^1\text{H}$  1.93 and  $^{13}\text{C}$  28.5 ppm),  $\gamma$ -CH ( $^1\text{H}$  1.73 ppm and  $^{13}\text{C}$  24.9), and  $\delta$ - $\text{CH}_2$  ( $^1\text{H}$  3.26 and  $^{13}\text{C}$  41.4 ppm) signals were assigned by means of  $^1\text{H}$ - $^1\text{H}$  TOCSY and  $^1\text{H}$ - $^{13}\text{C}$  HSQC experiments.

Arginine turned out to be the most abundant amino acid, as nitrogen reserve stored in the roots to ensure plant growth during germination and budbreak (Gomez, Vercambre, & Jordan, 2020; Keller & Senula, 2013; Yang et al., 2017).

**3.1.2.2. Organosulphur compounds.** As reported in literature (Putnik et al., 2019; Block, 1992) garlic bulbils, contain major organosulphur compounds (OSCs), namely  $\gamma$ -glutamylcysteines, non-protein amino acids, and their corresponding sulfoxide derivatives, being (+)-S-(2-propenyl)-L-cysteine sulfoxide (alliin), (+)-S-(trans-1-propenyl)-L-cysteine sulfoxide (isoalliin), and (+)-S-methyl-L-cysteine sulfoxide (methiin), (1S,3R,5S)-5-methyl-1,4-thiazane-3-carboxylic acid 1-oxide (cycloalliin), and S-(Prop-2-en-1-yl) prop-2-ene-1-sulfinothioate (allicin) the principal ones (Borlinghaus et al., 2014).

The  $^1\text{H}$  NMR spectrum of bulbils extracts show the presence of four organosulphur compounds namely allicin, methiin, and alliin and cycloalliin, see Fig. 3. In order to simplify the discussion about the assignment, the numbering of allicin will not adhere to the IUPAC standard. This is to ensure that the identical part of the molecule between allicin and alliin has the same numbering. According to the IUPAC numbering rules, the allyl-sulfoxide group should have priority. However, as shown in Fig. 3, the C atoms in allicin have been numbered starting from the allyl-sulfinio group. Here for the first time, it is reported the assignment of methiin, and alliin in the  $^1\text{H}$  spectrum of aerial bulbil garlic extracts. Methiin naturally occurring in the genus *Allium*, with

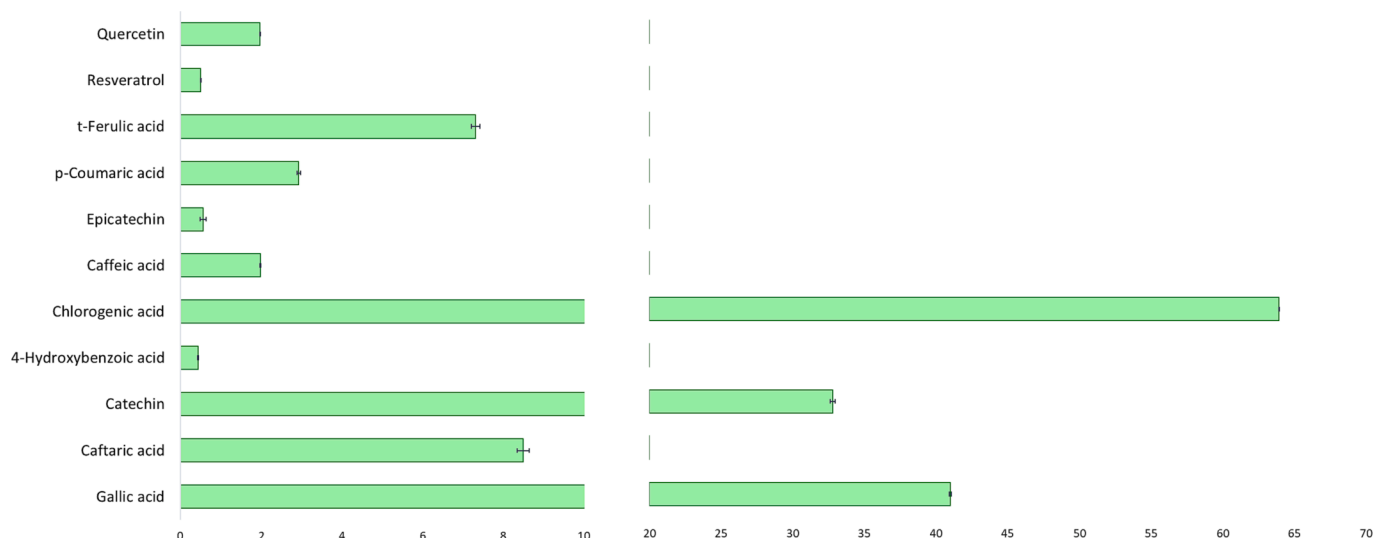


Fig. 4A. Histograms relative to concentration (mg/100 g) of phenolic compounds in the red garlic aerial bulbil hydroalcoholic extract. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

regards to *sativum* species (Rose et al., 2005), and some members of Brassicaceae, like *Brassica oleracea* L. (Ingallina et al., 2023) was identified by means of the spin correction observed in the  $^1\text{H}$ - $^1\text{H}$  TOCSY map between the  $\alpha$ -CH ( $^1\text{H}$  4.18 ppm,  $^{13}\text{C}$  59.3) and diastereotopic protons of  $\beta$ -CH<sub>2</sub> ( $^1\text{H}$  3.27, 3.47 ppm,  $^{13}\text{C}$  54.0 ppm). The assignment of  $\gamma$ -CH<sub>3</sub> group linked to the -S = O group 2.83 ppm ( $^{13}\text{C}$  39.1 ppm) was confirmed by literature data regarding other vegetable matrices (Saviano et al., 2019).

With regards to alliin and alliin, the two major allyl-sulphur metabolites, signals arising from allyl-organosulphur compounds were observed in spectral region between 5.00 and 6.00 ppm of the  $^1\text{H}$  NMR spectrum. In particular, the multiplet at 5.94 ppm ( $^{13}\text{C}$  125.5 ppm) assigned to the proton of CH-5 showed in the  $^1\text{H}$ - $^1\text{H}$  TOCSY map a spin correlation with: the protons of methylene CH<sub>2</sub>-6 at 5.57 ppm (dq  $J_{\text{H-H}} = 1.2$  Hz and  $J_{\text{H-Hcis}} = 10.2$  Hz) and 5.51 ppm (dq  $J_{\text{H-H}} = 1.2$  Hz and  $J_{\text{H-Htrans}} = 17.0$  Hz), and the diastereotopic methylene protons next to -S = O group at 3.64 and 3.87 ppm ( $^{13}\text{C}$  55.9 ppm) accounted for the CH<sub>2</sub>-4. The second spin system of alliin is analogue to the methiin. However, the  $^1\text{H}$ - $^{13}\text{C}$  HSQC spectrum showed a short-range correlation between the signals of diastereotopic methylene protons at 3.22 and 3.44 ppm and the corresponding carbon at 50.9 ppm, which accounted for the CH<sub>2</sub>-3. These signals belong, most likely, to alliin, as confirmed by comparison with the  $^1\text{H}$ ,  $^{13}\text{C}$  and  $^1\text{H}$ - $^{13}\text{C}$  HSQC spectra of the reference standard and literature data (Hakamata et al., 2015). The presence of alliin was assessed by the evidence of a further spin system consisting of the double

bond methines at 5.17 (dq  $J_{\text{H-H}} = 1.6$  Hz and  $J_{\text{H-Hcis}} = 10.0$  Hz), 5.21 (dq  $J_{\text{H-H}} = 1.6$  Hz and  $J_{\text{H-Htrans}} = 18.6$  Hz) and a multiplet at 5.84 ppm, corresponding to CH<sub>2</sub>-1 and CH-2, respectively. These signals showed a  $^1\text{H}$ - $^{13}\text{C}$  HSQC short-range correlation with the carbon at 118.9 and 134.7 ppm, respectively, and a  $^1\text{H}$ - $^1\text{H}$  TOCSY spin-correlation with methylene protons S-CH<sub>2</sub>-3 ( $^1\text{H}$  3.24 and  $^{13}\text{C}$  41.3 ppm). Signals belonging to the second part of the molecule overlapped in the  $^1\text{H}$  spectrum with the signals from alliin, and their assignment was based on 2D experiment and the literature data (Ritota et al., 2012; Liang et al., 2015).

Cycloalliin assignment was based on the presence of the characteristic doublet signal at 1.44 ppm ( $J_{\text{H-H}} = 6.5$  Hz) from the protons of the -methyl substituent in C-5, which showed a short-range correlation in the  $^1\text{H}$ - $^{13}\text{C}$  HSQC spectrum with the carbon at 19.5 ppm. Moreover, in the  $^1\text{H}$ - $^1\text{H}$  TOCSY spectrum, cross-peak correlations between the methyl substituent in C-5 and the proton at 3.93 ppm ( $^{13}\text{C}$  43.7 ppm), which accounted for the  $\alpha$  proton CH-5, and protons at 2.81 and 3.23 ppm, assigned to the methylene CH<sub>2</sub>-6 ( $^{13}\text{C}$  47.7 ppm) next to sulfoxide group were observed. These findings agreed with the literature data (Liang et al., 2015). The absence of isoalliin in the  $^1\text{H}$  NMR spectrum could be due to its exclusive biosynthesis in species of *Allium cepa*. (Borlinghaus et al., 2014). All the organosulphur compounds were identified and quantified (Fig. 2D), with the exception of alliin. This is due to the significant overlap of its characteristic signals with those of alliin and methiin.

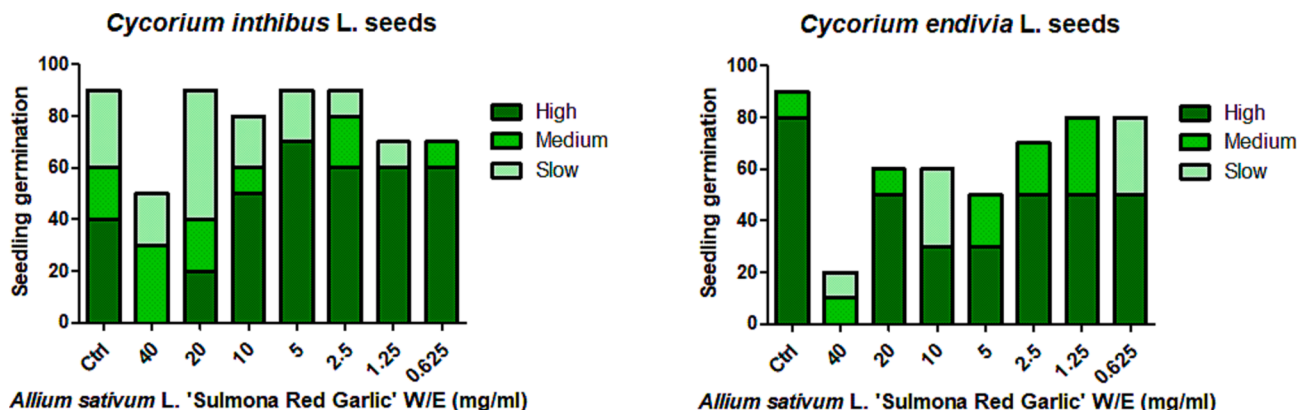


Fig. 5. Effects of the hydroalcoholic extract of aerial bulbils from "Sulmona Red Garlic" on the seedling germination of *Cycorium inthibus* and *Cycorium endivia*.



**Table 3**

Content in metabolites ( $\mu\text{g/mL}$ ) of the red garlic aerial bulbil hydroalcoholic extract. All identified phytochemicals have been identified through comparison with pure standards. Quantitative determination of the compounds was performed via DAD detector at 232–372 nm wavelength; nq: not quantified; nd: not detected.

Phytochemical	Rt	Wavelength	Concentration
1 Gallic acid	8.80	271	$16.400 \pm 0.059$
2 3-Hydroxytyrosol	11.71	275	nq
3 Caftaric acid	12.93	310	$3.397 \pm 0.147$
4 Catechin	14.80	278	$13.118 \pm 0.163$
5 4-Hydroxybenzoic acid	16.20	256	$0.176 \pm 0.012$
6 Chlorogenic acid	16.81	325	$25.573 \pm 0.010$
7 Vanillic acid	18.60	257	nq
8 Caffeic acid	19.00	325	$0.790 \pm 0.011$
9 Epicatechin	19.41	278	$0.226 \pm 0.078$
10 Syringic acid	20.05	274	nd
11 Syringaldehyde	21.80	310	nq
12 p-Coumaric acid	23.06	310	$1.170 \pm 0.045$
13 t-Ferulic acid	24.00	315	$2.925 \pm 0.105$
14 Hyperoside	26.92	254	nq
15 Rutin	27.16	254	nq
16 Resveratrol	27.70	310	$0.202 \pm 0.002$
17 t-Cinnamic acid	34.39	279	nq
18 Quercetin	35.89	372	$0.785 \pm 0.006$
19 Kaempferol	41.74	330	nq
20 Carvacrol	44.69	275	nd
21 Flavone	45.60	340	nq
22 3-Hydroxyflavone	46.05	340	nq
23 Emodin	47.70	289	nq

### 3.1.3. Miscellaneous

$^1\text{H}$  NMR spectrum revealed the presence of other metabolites reported in Table 1. According to Liang et al (Liang et al., 2015), pyroglutamic acid was identified due to the resonance of the diastereotopic protons  $\beta$ ,  $\beta'$ - $\text{CH}_2$  at 2.04 and 2.51 ppm, respectively ( $^{13}\text{C}$  26.5 ppm). This signal showed in the  $^1\text{H}$ - $^1\text{H}$  TOCSY map a spin correlation with  $\gamma$ - $\text{CH}_2$  protons at 2.40 ppm ( $^{13}\text{C}$  32.5 ppm). Resonance at 9.13 ppm showed a  $^1\text{H}$ - $^1\text{H}$  TOCSY cross-peak with proton at 8.84 ppm and was assigned to trigonelline, in accordance with data reported in literature

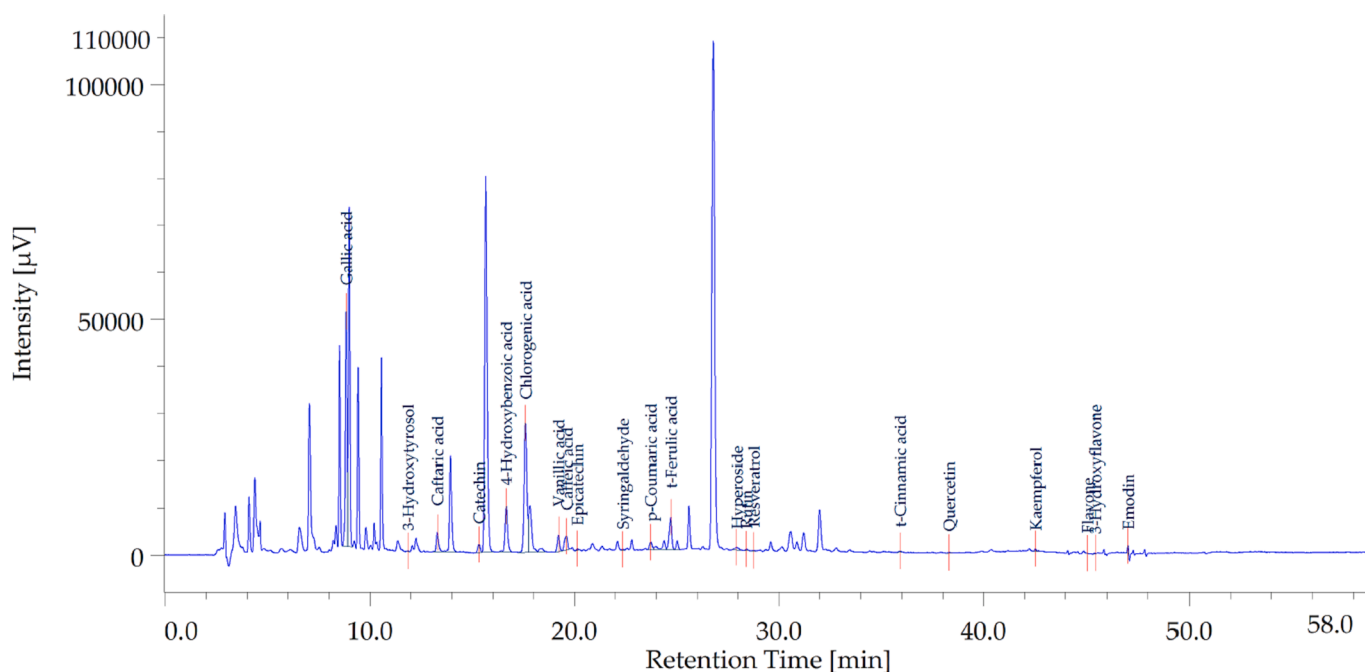
(Ritota et al., 2012). Singlet at 3.27 ppm showing a correlation in  $^1\text{H}$ - $^1\text{H}$  TOCSY with proton at 3.81 ppm ( $\alpha$ - $\text{CH}_2$ ) was attributed to the +  $\text{N}(\text{CH}_3)_3$  methyl groups of glycine betaine. Its presence was also confirmed by the  $^1\text{H}$ - $^{13}\text{C}$  HSQC spectrum, which yield correlations between each proton with their carbons at 54.4 and 67.3 ppm, respectively. Also, tyramine was identified based on the presence of the characteristic doublet at 6.90 ppm assigned to CH-2,6. This signal showed in the  $^1\text{H}$ - $^1\text{H}$  TOCSY experiment a correlation with protons CH-3,5 at 7.21 ppm. The  $^1\text{H}$ - $^{13}\text{C}$  HSQC allowed the short-range correlation between these protons with corresponding carbons at 116.6 and 131.6 ppm, respectively. Finally, according to literature, choline and uridine were identified in the hydroalcoholic extract of Sulmona Red Garlic aerial bulbils. In detail, singlet at 3.20 ppm was attributed to choline, whereas the presence of uridine was assessed by the presence of the doublet at 7.88 ( $J_{\text{H-H}} = 7.9$  Hz), assigned to CH-6, which correlated in the  $^1\text{H}$ - $^1\text{H}$  TOCSY spectrum with the CH-1' proton at 5.90 ppm (Fig. 2E).

### 3.2. Phenolic compounds by HPLC-DAD analysis

The hydroalcoholic extract of garlic aerial bulbils was also assayed through HPLC-DAD to obtain the phenolic quantification. The experimental data are reported in Figs. 4A and Fig. 4B and Table 3 that showed gallic acid, catechin, and chlorogenic acid as the prominent phytochemicals. These compounds were present at concentrations ranging from 13.118 to 25.573  $\mu\text{g/mL}$ , and this is consistent, albeit partially, with literature (Recinella et al., 2022). On the other hand, t-ferulic and caftaric acid ranged from 2.925 and 3.397 ( $\mu\text{g/mL}$ ), respectively. Collectively, the extract's concentration of the present phytochemicals is consistent with potential antioxidant and anti-inflammatory effects (Chiavaroli et al., 2021; Recinella et al., 2022); thus, further supporting the pharmacological evaluation described below.

### 3.3. Eco-toxicological studies

In order to validate the bio-pharmacological use of the hydroalcoholic extract from the aerial bulbils of the "Sulmona Red Garlic" ecotype, a multidirectional ecotoxicological approach was conducted in



**Fig. 4B.** Chromatographic analysis of the phenolic compounds present in the hydroalcoholic extract from the aerial bulbils of "Sulmona Red Garlic". Among the identified phytochemicals, the prominent were: gallic acid, caftaric acid, catechin, chlorogenic acid, and t-ferulic acid. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

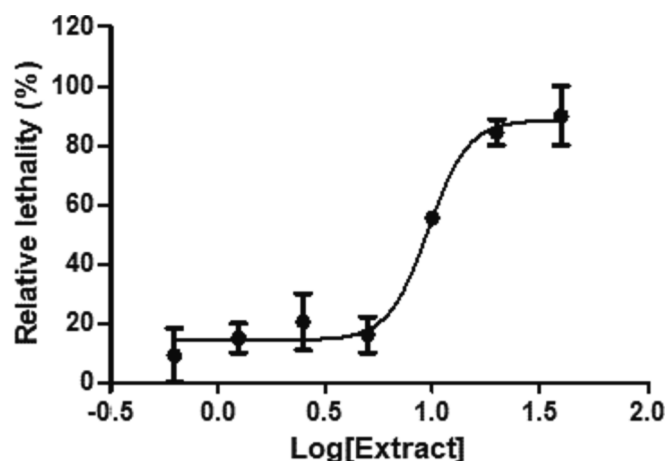


Fig. 6. Effect of the hydroalcoholic extract of aerial bulbils from “Sulmona Red Garlic” on the viability of the *Artemia salina* (Brine shrimp lethality assay).  $LC_{50}$ : 9.581 mg/mL.

order to discriminate the limits of biocompatibility in eukaryotic organisms. In this regard, a former investigation was carried out to study the phytotoxicity of the extract against two edible and herbaceous dicotyledon species, namely *Cichorium intybus* and *C. endivia*. The seedling germination was evaluated for both of them, after exposing the seeds to the extract (0.625–40 mg/mL). The extract caused a significant reduction (over 50 %) of the seedling germination at the highest tested concentration that was considered phytotoxic (Fig. 5). This is consistent, albeit partially, with the extract content in phenolic compounds that are well-known to work as allelochemicals (Dehghanian et al., 2022). In parallel, the effects of the extract on the survival of the *Artemia salina* (brine shrimp) was evaluated, as well. In this context, the extract was tested in the same concentration range (0.625–40 mg/mL). As already observed for polar extracts prepared with ultrasound-assisted maceration from other edible plants (Chiavaroli et al., 2021; Orlando et al., 2021) a  $LC_{50}$  value  $\approx$  10 mg/mL was provided by the brine shrimp lethality test (Fig. 6). This  $LC_{50}$  value was therefore chosen as concentration limit for the *Daphnia magna* toxicity assay. In this in vitro model,

the extract did not influence the heart rate of the *D. magna* in both basal and ethanol-induced toxicity conditions (Fig. 7); thus, suggesting a different sensitivity of the *D. magna* to the extract, compared with *A. salina*. In agreement with previous studies (Chiavaroli et al., 2021; Orlando et al., 2021), a concentration at least ten-fold lower than the  $LC_{50}$  calculated with brine shrimp lethality assay was chosen as putative biocompatibility limit for the in vitro assays described below.

### 3.4. In vitro studies

The garlic extract (200–1000  $\mu$ g/mL) was also tested on colon cancer HCT116 cell line, for studying cytotoxicity properties. In parallel, the extract was assayed on the myocyte C2C12 cell line, chosen as non-tumoral comparison cell line. Intriguingly, the two cell lines showed a different sensitivity to the extract; indeed, C2C12 cells tolerated the extract up to the concentration of 1000  $\mu$ g/mL (Fig. 8A); thus, demonstrating a good biocompatibility as expected from the abovementioned eco-toxicological models. By contrast, the extract determined a significant and concentration-dependent reduction of HCT116 cell viability (Fig. 8B-C), starting from the concentration of 200  $\mu$ g/mL, that adds, albeit partially, to recent in vitro and in vivo findings (Ansary et al., 2020; Recinella et al., 2022; Tanrikulu et al., 2020).

In HCT116 cells, the gene expression of different factors involved in angiogenesis, cell migration, and inflammation-to-cancer transition, namely  $TNF\alpha$ ,  $HIF1\alpha$ , and VEGFA (Chen et al., 2019; Panyatthep, Punturee, & Chewonarin, 2022) was assayed as well, in both basal and hydrogen peroxide-induced oxidative stress conditions. The hydrogen peroxide stimulus was chosen for its capability to increase the gene expression of the abovementioned factors, in colon cancer cells (Chiavaroli et al., 2023; Recinella et al., 2022). Particularly, the reduction of the gene expression of  $TNF\alpha$ ,  $HIF1\alpha$ , and VEGFA (Fig. 9A-C) indicates the capability of the extract in contrasting cancer cell proliferation and angiogenesis (Kim et al., 2015; Wei et al., 2023; Zhang & Luo, 2018). The inhibition of  $TNF\alpha$  gene expression is also consistent with the anti-inflammatory effects demonstrated by garlic extracts in the colon (Recinella et al., 2022). A new field of research is also considering the involvement of TRPM8 in colon carcinogenesis (Borrelli et al., 2014; Liu, Li, & Xu, 2022). Additionally, recent findings of ours suggested a relationship between cytotoxic effect and reduced TRPM8 gene expression,

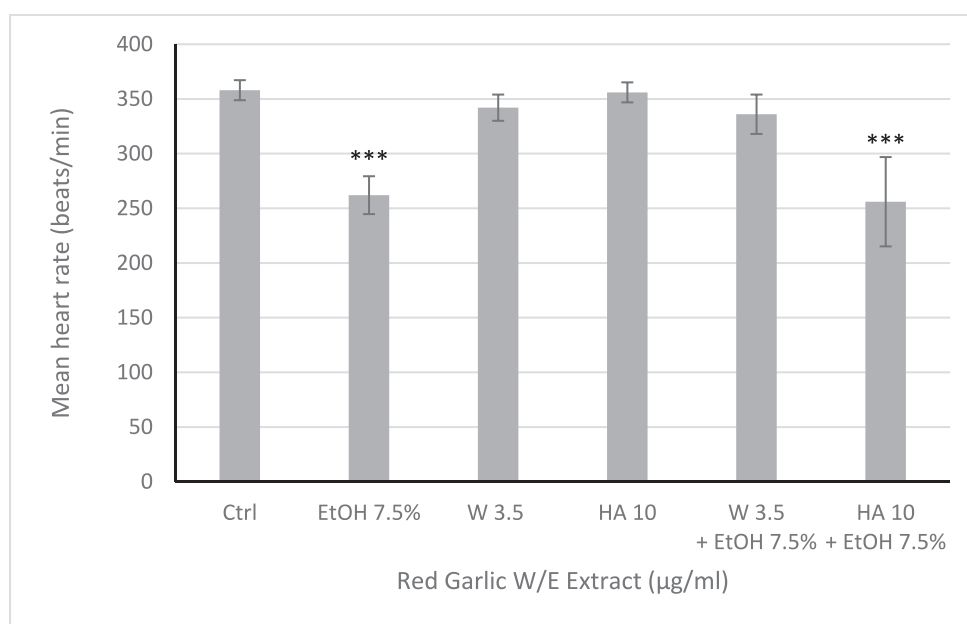
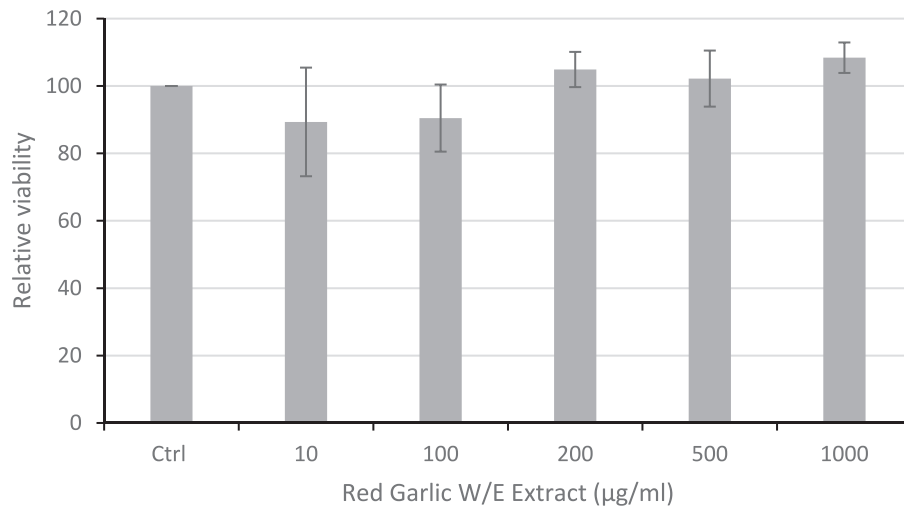
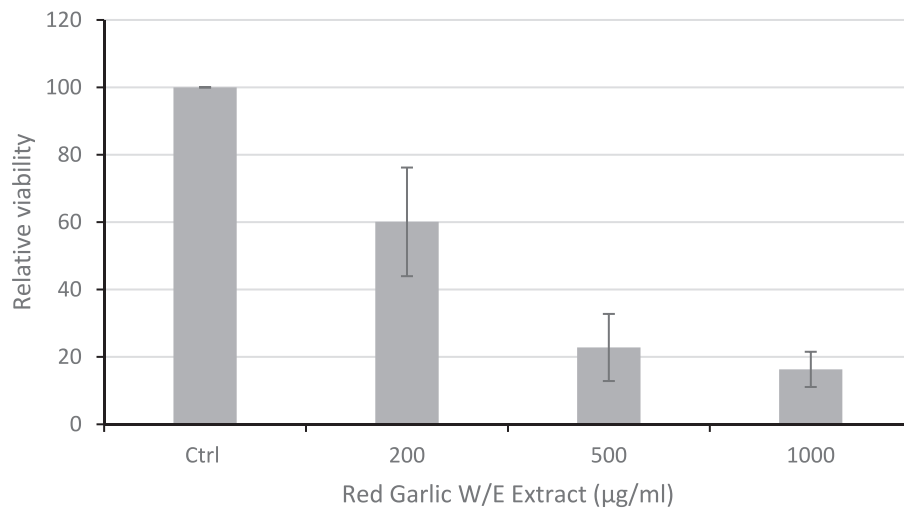


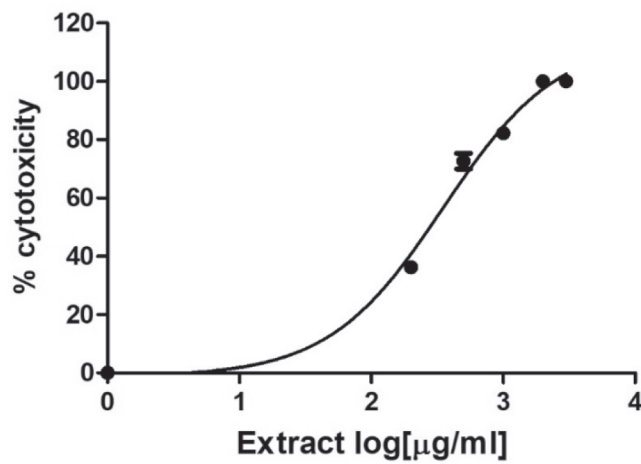
Fig. 7. Null effect of the hydroalcoholic (W/E) extract of aerial bulbils from “Sulmona Red Garlic” on the heart rate of *Daphnia magna* (*Daphnia magna* heart rate test). ANOVA,  $P < 0.0001$ ,  $***P < 0.001$  vs. Ctrl (Control group).



(A)

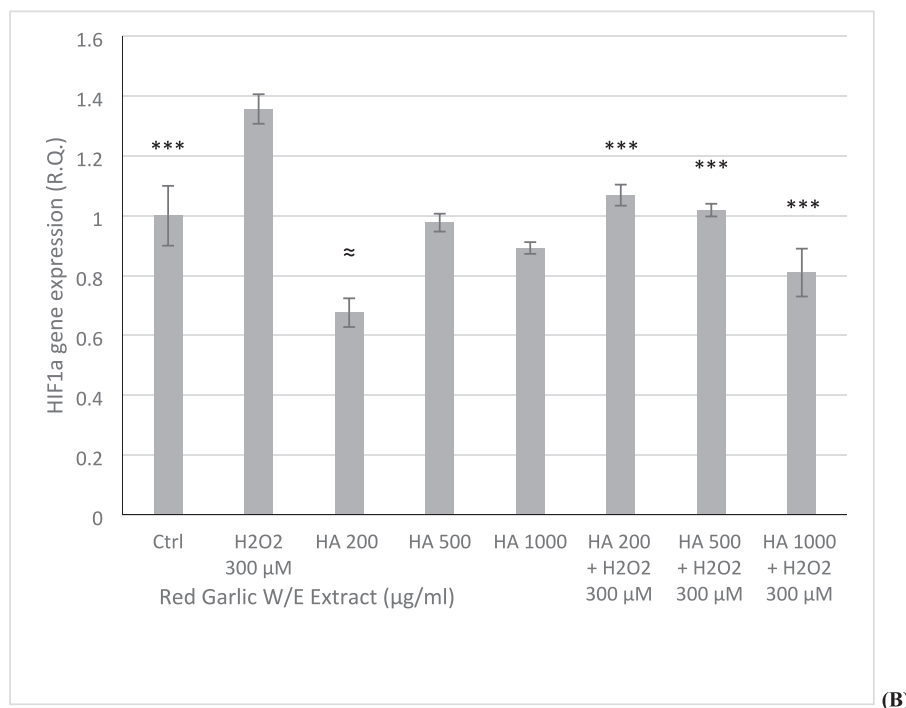
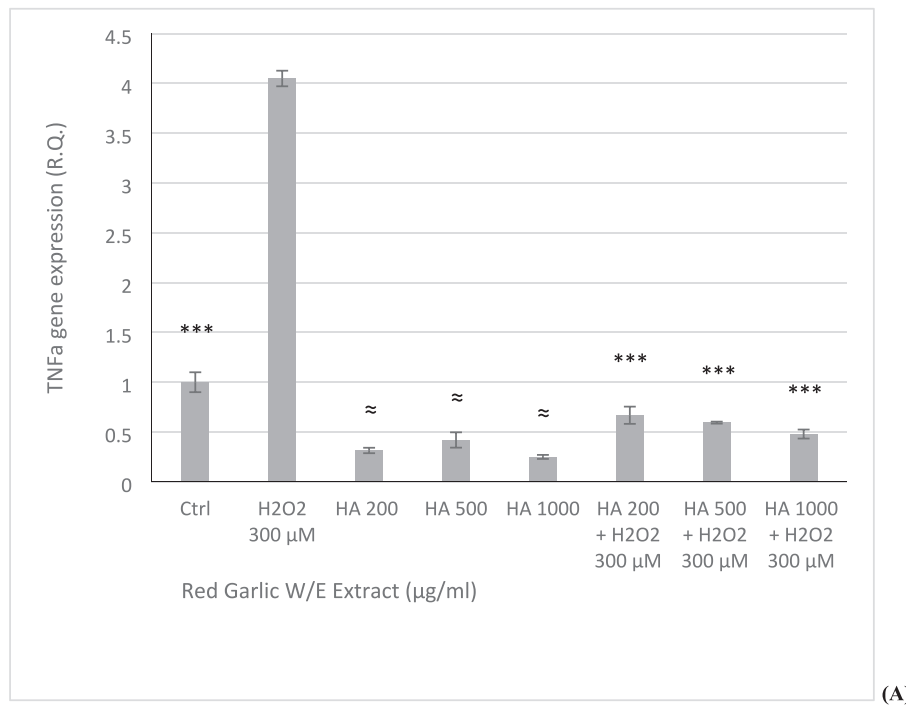


(B)

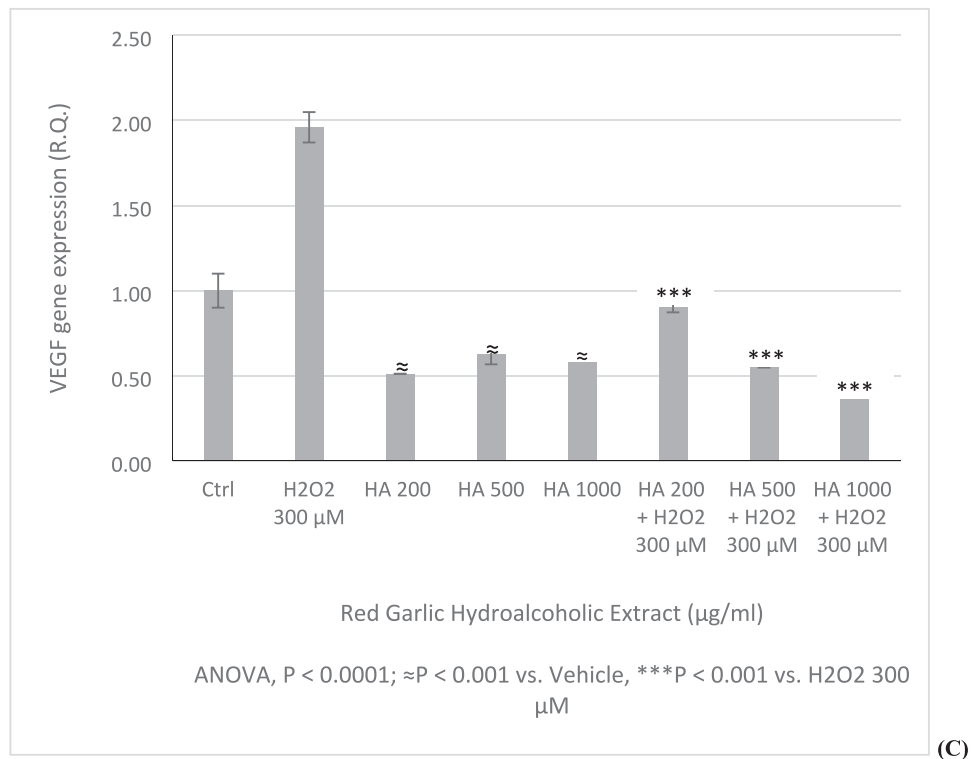


(C)

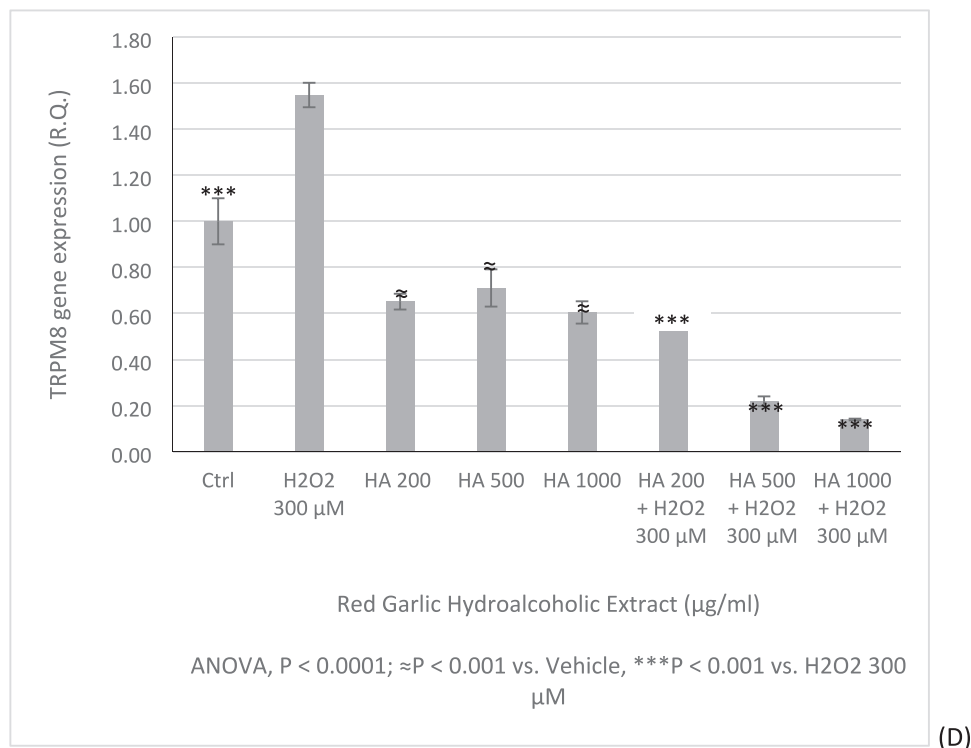
**Fig. 8.** Effects of the hydroalcoholic (W/E) extract of aerial bulbils from “Sulmona Red Garlic” on the C2C12 and HCT116 (B) cell line viability. The extract exerted a cytotoxic effect only on the human colon cancer HCT116 cell line (C: IC<sub>50</sub>: 352.07 µg/mL). ANOVA, P < 0.0001; \*\*\*P < 0.001 vs. Ctrl (Control group).



**Fig. 9.** Inhibitory effect induced by the hydroalcoholic extract of aerial bulbils from “Sulmona Red Garlic” on basal and hydrogen peroxide induced TNFα (A), HIF1α (B), VEGFA (C), and TRPM8 (C) gene expression, in HCT116 cells. A: ANOVA,  $P < 0.0001$ ; \*\*\* $P < 0.001$  vs. hydrogen peroxide (H.P.);  $\approx P < 0.0001$  vs Ctrl (Control group). B: ANOVA,  $P < 0.0001$ ; \*\*\* $P < 0.001$  vs. hydrogen peroxide (H.P.);  $\approx P < 0.0001$  vs Ctrl (Control group). C: ANOVA,  $P < 0.0001$ ; \*\*\* $P < 0.001$  vs. hydrogen peroxide (H.P.);  $\approx P < 0.0001$  vs Ctrl (Control group). D: ANOVA,  $P < 0.0001$ ; \*\*\* $P < 0.001$  vs. hydrogen peroxide (H.P.);  $\approx P < 0.0001$  vs Ctrl (Control group). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)



(C)

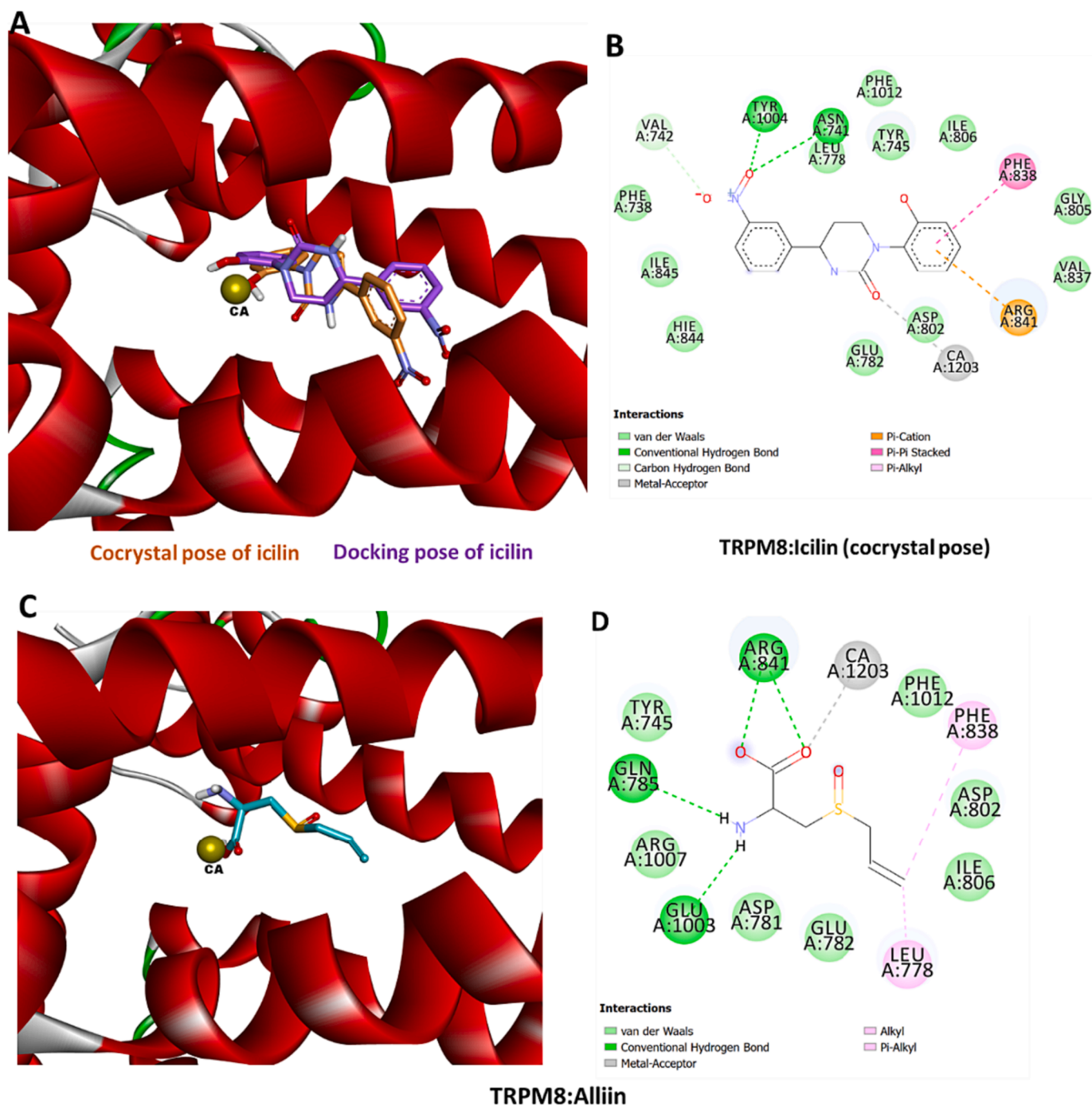


(D)

Fig. 9. (continued).

in colon cancer cells, in vitro (Chiavaroli et al., 2023; Recinella et al., 2022). In the present study, the hydroalcoholic extract of aerial bulbils form “Sulmona Red Garlic” reduced the gene expression of TRPM8 (Fig. 9D); thus, pointing to the inhibition of TRPM8 gene expression as

one of the possible mechanisms underlying the extract-induced cytotoxic effect against HCT116 cells. The gene expression inhibition of all tested proteins could be related, albeit partially, to the extract’s content in phenolic compounds, among which catechin demonstrated a direct



**Fig. 10.** (A) Validation of docking method: superimposition of the cocrystal pose of icilin in the cryo-EM structure of the TRPM8 ion channel complexed with icilin (PDB ID: 6NR3) with the docking pose (B) interaction between icilin and TRPM8. (C) Binding pose of alliin in the binding pocket of TRPM8, and (D) interaction between alliin and TRPM8.

inhibitory effect on TRPM8 gene expression (Recinella et al., 2022).

### 3.5. *In silico* studies

Bioinformatics predictions conducted on the platform STITCH also suggested that alliin and allicin could play a pivotal role in mediating extract-induced cytotoxic effect, as demonstrated by their putative interactions with HIF1 signaling pathway (pathway ID 04066), TNF signaling pathway (pathway ID 04668), VEGF signaling pathway (pathway ID 04370), and TRP channels pathway (pathway ID 04750).

Regarding the TRPM8, a docking approach was also conducted to

calculate the putative affinities of both alliin and allicin towards this endovanilloid receptor. To validate the binding-mode predictive ability of AutoDock 4.2.6, the cocrystal ligand icilin was redocked into its cognate protein TRPM8. The cocrystal and docking poses of icilin were superimposed in the binding pocket of TRPM8 with the root-mean-square displacement (RMSD) value of 1.74 Å (Fig. 10A). An RMSD value of < 2.00 Å suggests that a docking pose is well predicted (Uba and Yelekçi, 2019). In the cocrystal pose, icilin bound to the TRPM8 via variety of interactions, including metal-acceptor interaction between the  $Ca^{2+}$  ion and oxo substituent; H-bonds with Asn71 and Tyr1004. Other interactions comprise  $\pi$ -cation,  $\pi$ - $\pi$  stacked, and multiple van der

Waals interactions all over the binding pocket (Fig. 10B). Similarly, alliin was completely buried in the binding channel of TRPM8 (Fig. 10C). Alliin engaged Ca<sup>2+</sup> ion in metal-acceptor interaction via its carbonyl oxygen, formed H-bonds with Gln785, Arg841, and Glu1003; hydrophobic interactions with Leu778 and Phe838; and multiple van der Waals interactions with the residues lining the TRPM8 pocket (Fig. 10D). These interactions yielded a docking binding energy score of −6.84 kcal/mol, which is very much comparable with the docking score of the cocystal ligand, icilin (−6.77 kcal/mol). Thus, alliin likely exerts its activity by interacting with TRPM8. An acceptable affinity for the receptor was calculated for only alliin (9.83 μM), and this is consistent with possible direct interactions between the compound and the receptor. Whilst allicin showed a putative affinity > 100 μM (data not shown); thus, ruling out any direct interaction between allicin and TRPM8. By contrast, allicin demonstrated to activate endovanilloid channels TRPV1 and TRPA1, thus indicating analgesic properties of this compounds (Hernández-Cruz et al., 2022; Xiao et al., 2021).

The ethanolic extract (EtOH:H<sub>2</sub>O 1:1 v/v) of the red garlic aerial bulbils, submitted to biological evaluation, was also analysed by NMR spectroscopy to assess the presence of alliin and other organosulphur compounds. The results clearly showed that the metabolomic profile was consistent with that of the hydroalcoholic phase of the Bligh-Dyer extract, which included OSCs such as allicin, methiin, alliin and cycloalliin, to which the activity on HCT116 cells and the affinity towards TRPM8 receptor are attributable.

#### 4. Conclusion

In the present study, the composition and biological activity of the hydroalcoholic extract from the aerial bulbils of the *A. sativum* ecotype “Sulmona Red Garlic” was investigated. The extract was analyzed using a multi-methodological approach. The untargeted NMR spectroscopy provided valuable insights regarding several classes of metabolites in the hydroalcoholic extract of red garlic aerial bulbils. This includes organosulphur compounds such as alliin, allicin, methiin, and cycloalliin, all of which were quantified except for alliin. Targeted analysis allowed to identify and quantify secondary metabolites belonging to polyphenols. Aerial bulbils showed a chemical profile rich in healthy compounds to which the biological effect of the extract is attributable, confirming their promising phytotherapeutic use. The extract was effective in inducing a cytotoxic effect against human colon cancer HCT116 cells, in the concentration range 200–1000 μg/mL. In parallel, the gene expression of TNFα, HIF1α, VEGFA, and TRPM8 was reduced. Both sulphur and phenolic compounds could be responsible, albeit partially, of such effects. Intriguingly, the extract was well-tolerated by the non-tumoral C2C12 cell line, in the same concentration-range; thus, further suggesting the biocompatibility of the tested extract.

Concluding, the present study demonstrated the potential application of a plant material, the aerial bulbils, that is currently considered as waste material. However, the careful and manual work selection of harvested garlic plants makes the aerial bulbils an innovative and high-quality by-product, with promising health-promoting applications, as demonstrated by the metabolomics and biological data. Indeed, aerial bulbils could be considered as secondary material on which to base the development of innovative products such as food supplements with protective effects in the colon. This could lead to an overall improvement of the whole “Sulmona Red Garlic” productive chain.

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D93C22000890001, Project title “ON Foods – Research and innovation network on food and nutrition Sustainability, Safety and Security – Working ON Foods”). ]

#### Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

#### Data availability

Data will be made available on request.

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#### Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.foodres.2023.113654>.

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